$O(17)$ and $N(1)$. This interaction does not result in any severe changes in torsion angles.

The heterocyclic $D$ ring is cis-fused to ring $C$ and 1,3-diaxially fused to ring $E$. Alternatively, it can be considered to be a component of a bridged eightmembered ring. There are several opposing interactions which determine the torsion angles in the heterocyclic six-membered ring. The fusion of the $C(4 a)$ and $C(3)$ axial substituents tends to increase the torsion angles while interactions $\mathrm{H}(4 \mathrm{r}) \cdots \mathrm{H}(15 \gamma)=$ $2 \cdot 04, \mathrm{H}(11 \mathrm{a}) \cdots \mathrm{H}(15 \gamma)=2.33$ and $\mathrm{H}(12 \beta) \cdots \mathrm{N}(1)=$ $2.72 \AA$ tend to flatten the ring and reduce the torsion angles.

Ring $E$ is cis-fused to ring $C$ and 1,3-diaxially fused to ring $D$. The diaxial fusion and interactions $\mathrm{H}(12 \alpha) \cdots \mathrm{H}(16 \gamma)=2.51, \mathrm{H}(12 \beta) \cdots \mathrm{N}(1)=2.72$ and $\mathrm{H}(13) \cdots \mathrm{H}(14 \beta)=2.23 \AA$ determine the variations in torsion angles. The eight-membered ring assumes a boat conformation. The H on $\mathrm{O}(17)$ is bonded intramolecularly to $\mathrm{N}(1)$ with $\mathrm{N}(1) \cdots \mathrm{O}(17)=2.763$, $\mathrm{N}(1) \cdots \mathrm{H}(17)=2.11$ and $\mathrm{H}(17) \cdots \mathrm{O}(17)=0.87 \AA$. $\mathrm{O}(18)$ is intermolecularly hydrogen bonded to $\mathrm{N}(6)$ with $\mathrm{N}(6) \cdots \mathrm{O}(18)=2 \cdot 891, \mathrm{~N}(6) \cdots \mathrm{H}(18)=2 \cdot 18$ and $\mathrm{H}(18) \cdots \mathrm{O}(18)=0.87 \AA$.

Aristotelinine can be considered a biological oxidation product of aristoteline. These alkaloids represent
a unique structural type which may be restricted to the genus Aristotelia.

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## References

Anderson, B. F., Robertson, G., Avey, H. P., Donovan, W. F., Bick, I. R. C., Bremner, J. B., Finney, A. J. T., Preston, N. W., Gallagher, R. T. \& Russell, G. R. (1975). J. Chem. Soc. Chem. Commun. pp. 511-513.

Bittner, M., Silva, M., Gopalakrishna, E. M., Watson, W. H., Zabel, V., Matlin, S. A. \& Sammes, P. G. (1978). J. Chem. Soc. Chem. Commun. 79-80.

Bucourt, R. (1974). Topics in Stereochemistry, Vol. 8, pp. 159-224, edited by E. L. Eliel \& N. L. Allinger. New York: Wiley-Interscience.
Germain, G., Main, P. \& Woolfson, M. M. (1971). Acta Cryst. A27, 368-376.
International Tables for X-ray Crystallography (1974). Vol. IV, pp. 71-98. Birmingham: Kynoch Press.
Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee.

Acta Cryst. (1978). B34, 3780-3781

# 2,3-Dihydro-2-(4-methoxyphenyl)-3-phenyl-4H-1,3,2-benzoxazaphosphorine-4-thione 2-Oxide 

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#### Abstract

C}_{20} \mathrm{H}_{16} \mathrm{NO}_{3} \mathrm{PS}, M_{r}=381 \cdot 4\), monoclinic, $P 2_{1} / c, a=12.96(2), b=15.60(2), c=9.49$ (1) $\AA$, $\beta=103.8(2)^{\circ}, V=1863.3 \AA^{3}, Z=4, D_{c}=1.36$ $\mathrm{g} \mathrm{cm}^{-3}$. Refinement gave $R=0.056$ and $R_{w}=0.060$ for 235 parameters and 1761 significant reflexions [ $I>3 \sigma(I)$ ]. The stereochemistry of the molecule has been established.

Introduction. The title compound was prepared by Scheibye, Pedersen \& Lawesson (1978a,b) as yellow crystals (m.p. $173^{\circ} \mathrm{C}$ ). The crystals were kindly provided by $S$. Scheibye.

A crystal $1.0 \times 0.7 \times 0.2 \mathrm{~mm}$ was mounted about c. The lattice type and space group were established from photographs taken with Cu and Mo radiations.




Fig. 1. A stereoscopic drawing of the molecule showing the numbering of the atoms.

Table 1. Fractional atomic coordinates $\left(\times 10^{4}\right)$ and their e.s.d.'s

|  | $x$ | $y$ | $z$ |
| :--- | ---: | ---: | ---: |
|  |  |  |  |
| S | $5117(2)$ | $437(1)$ | $2642(2)$ |
| P | $8131(1)$ | $840(1)$ | $5527(2)$ |
| $\mathrm{O}(1)$ | $7659(4)$ | $1649(3)$ | $6195(5)$ |
| $\mathrm{O}(2)$ | $8902(4)$ | $1033(3)$ | $4687(5)$ |
| $\mathrm{O}(3)$ | $9426(4)$ | $-1490(3)$ | $10506(5)$ |
| N | $698(4)$ | $409(3)$ | $4490(5)$ |
| $\mathrm{C}(1)$ | $6116(5)$ | $882(4)$ | $3803(7)$ |
| $\mathrm{C}(2)$ | $6155(6)$ | $1806(4)$ | $4152(7)$ |
| $\mathrm{C}(3)$ | $6898(6)$ | $2161(4)$ | $5299(7)$ |
| $\mathrm{C}(4)$ | $6918(6)$ | $3019(5)$ | $5657(8)$ |
| $\mathrm{C}(5)$ | $6160(7)$ | $3553(5)$ | $4810(10)$ |
| $\mathrm{C}(6)$ | $5421(7)$ | $3222(5)$ | $3655(10)$ |
| $\mathrm{C}(7)$ | $5414(6)$ | $2368(5)$ | $3337(8)$ |
| $\mathrm{C}(8)$ | $7085(5)$ | $-496(4)$ | $4157(6)$ |
| $\mathrm{C}(9)$ | $6643(6)$ | $-1106(5)$ | $4859(7)$ |
| $\mathrm{C}(10)$ | $6782(7)$ | $-1971(5)$ | $4591(8)$ |
| $\mathrm{C}(11)$ | $7351(7)$ | $-2200(5)$ | $3617(9)$ |
| $\mathrm{C}(12)$ | $7784(6)$ | $-1590(6)$ | $2911(9)$ |
| $\mathrm{C}(13)$ | $7662(6)$ | $-731(5)$ | $3183(7)$ |
| $\mathrm{C}(14)$ | $8544(5)$ | $155(4)$ | $7030(6)$ |
| $\mathrm{C}(15)$ | $9428(5)$ | $-361(5)$ | $7150(7)$ |
| $\mathrm{C}(16)$ | $9745(5)$ | $-921(4)$ | $8280(7)$ |
| $\mathrm{C}(17)$ | $9183(5)$ | $-961(4)$ | $9347(7)$ |
| $\mathrm{C}(18)$ | $8293(6)$ | $-442(5)$ | $9240(7)$ |
| $\mathrm{C}(19)$ | $7977(5)$ | $111(5)$ | $8103(8)$ |
| $\mathrm{C}(20)$ | $10250(7)$ | $-2131(5)$ | $10575(9)$ |

Intensities were collected on a computer-controlled Supper diffractometer with graphite-monochromated Mo $K \kappa$ radiation, a scintillation counter and a pulseheight analyser. 4366 independent reflexions were measured by the background-peak-background method out to $\sin \theta=0.5 .1761$ significant reflexions with $I>3 \sigma(I)$ were used in the subsequent calculations. No correction was applied for absorption.
The structure was determined with MULTAN (Germain, Main \& Woolfson, 1971). Coordinates, anisotropic thermal parameters and a scale factor were refined by full-matrix least-squares calculations. The positions of 13 H atoms attached to the aromatic rings were calculated and included in the refinement but the geometrical and thermal parameters of the H atoms were not refined. The final $R=0.056$ and $R_{w}=0.060$ for 235 parameters and 1761 significant reflexions $[I>3 \sigma(I)]$. Atomic coordinates are listed in Table 1.*

Discussion. Scheibye, Pedersen \& Lawesson (1978a,b) showed that salicylanilide reacts with the dimer of $p$-methoxyphenylthioxophosphine sulphide to give the corresponding thioamide (yield $51 \%$ ), but in addition

[^0]Table 2. Bond lengths ( $\AA$ ) and their e.s.d.'s

| $\mathrm{P}-\mathrm{N}$ | $1.710(6)$ | $\mathrm{C}(8)-\mathrm{C}(9)$ | $1.363(9)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N}-\mathrm{C}(1)$ | $1.376(8)$ | $\mathrm{C}(9)-\mathrm{C}(10)$ | $1.394(10)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.477(9)$ | $\mathrm{C}(10)-\mathrm{C}(11)$ | $1.361(11)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.385(9)$ | $\mathrm{C}(11)-\mathrm{C}(12)$ | $1.360(12)$ |
| $\mathrm{C}(3)-\mathrm{O}(1)$ | $1.391(8)$ | $\mathrm{C}(12)-\mathrm{C}(13)$ | $1.382(10)$ |
| $\mathrm{O}(1)-\mathrm{P}$ | $1.599(5)$ | $\mathrm{C}(13)-\mathrm{C}(8)$ | $1.369(9)$ |
| $\mathrm{P}-\mathrm{O}(2)$ | $1.451(5)$ | $\mathrm{C}(14)-\mathrm{C}(15)$ | $1.383(9)$ |
| $\mathrm{P}-\mathrm{C}(14)$ | $1.760(7)$ | $\mathrm{C}(15)-\mathrm{C}(16)$ | $1.368(9)$ |
| $\mathrm{N}-\mathrm{C}(8)$ | $1.459(8)$ | $\mathrm{C}(16)-\mathrm{C}(17)$ | $1.383(10)$ |
| $\mathrm{C}(1)-\mathrm{S}$ | $1.641(7)$ | $\mathrm{C}(17)-\mathrm{C}(18)$ | $1.393(9)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.380(10)$ | $\mathrm{C}(18)-\mathrm{C}(19)$ | $1.365(10)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.388(11)$ | $\mathrm{C}(19)-\mathrm{C}(14)$ | $1.393(9)$ |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.373(11)$ | $\mathrm{C}(17)-\mathrm{O}(3)$ | $1.351(8)$ |
| $\mathrm{C}(6)-\mathrm{C}(7)$ | $1.365(11)$ | $\mathrm{O}(3)-\mathrm{C}(20)$ | $1.453(9)$ |
| $\mathrm{C}(7)-\mathrm{C}(2)$ | $1.391(10)$ |  |  |

Table 3. Angles $\left({ }^{\circ}\right)$ and their e.s.d.'s

| 1) $-\mathrm{P}-\mathrm{N}$ | $100 \cdot 2$ (3) | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | 120.7 (8) |
| :---: | :---: | :---: | :---: |
| $\mathrm{O}(1)-\mathrm{P}-\mathrm{O}(2)$ | 115.6 (3) | $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(2)$ | 121.7 (8) |
| $\mathrm{O}(1)-\mathrm{P}-\mathrm{C}(14)$ | 103.2 (3) | $\mathrm{N}-\mathrm{C}(8)-\mathrm{C}(9)$ | 120.0 (6) |
| $\mathrm{O}(2)-\mathrm{P}-\mathrm{N}$ | 112.5 (3) | $\mathrm{N}-\mathrm{C}(8)-\mathrm{C}(13)$ | 119.7 (6) |
| $\mathrm{O}(2)-\mathrm{P}-\mathrm{C}(14)$ | $117 \cdot 1$ (3) | C(9)-C(8)-C(13) | 120.3 (7) |
| N-P--C(14) | 106.4 (3) | $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | 119.9 (7) |
| $\mathrm{P}-\mathrm{N}-\mathrm{C}(1)$ | 124.2 (5) | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | 119.6 (7) |
| $\mathrm{P}-\mathrm{N}-\mathrm{C}(8)$ | 113.4 (5) | $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | 120.4 (8) |
| (1)-N-C(8) | 121.4 (6) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(13)$ | 120.5 (8) |
| C(1)-C(2) | 115.9 (6) | $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(8)$ | 119.4 (7) |
| -C(1)-S | 121.2 (5) | $\mathrm{P}-\mathrm{C}(14)-\mathrm{C}(15)$ | 120.4 (5) |
| (2)-C(1)-S | 122.9 (6) | $\mathrm{P}-\mathrm{C}(14)-\mathrm{C}(19)$ | 120.8 (5) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $123 \cdot 1$ (7) | $\mathrm{C}(15)-\mathrm{C}(14)-\mathrm{C}(19)$ | 118.9 (7) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(7)$ | 120.6 (7) | $\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)$ | 121.6 (7) |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(7)$ | 116.3 (7) | $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | 119.4 (7) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{O}(1)$ | 120.7 (7) | $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | 119.4 (7) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 123.4 (7) | $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{O}(3)$ | 124.3 (7) |
| $\mathrm{O}(1)-\mathrm{C}(3)-\mathrm{C}(4)$ | 115.9 (7) | $\mathrm{C}(18)-\mathrm{C}(17)-\mathrm{O}(3)$ | 116.3 (7) |
| $\mathrm{C}(3)-\mathrm{O}(1)-\mathrm{P}$ | 119.6 (4) | $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)$ | 120.8 (7) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 118.0 (8) | $\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(14)$ | 119.9 (7) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 120.0 (8) | $\mathrm{C}(17)-\mathrm{O}(3)-\mathrm{C}(20)$ | 118.9 (6) |

two P-containing heterocyclic systems were isolated namely $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{NO}_{3} \mathrm{PS}$ (yield $21 \%$, m.p. $173{ }^{\circ} \mathrm{C}$ ) and $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{NO}_{2} \mathrm{PS}_{2}$ (yield $8 \%$, m.p. $149^{\circ} \mathrm{C}$ ).

It was not possible to determine the structures of these P-containing heterocyclic compounds on the basis of analytical and spectral data and therefore the molecular structure of $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{NO}_{3} \mathrm{PS}$ was established by X-ray diffraction analysis.

Fig. 1 is a stereoscopic drawing of the molecule showing the numbering of the atoms. Bond distances are given in Table 2, and angles in Table 3.

## References

Germain, G., Main, P. \& Woolfson, M. M. (1971). Acta Cryst. A27, 368-376.
Scheibye, S., Pedersen, B. S. \& Lawesson, S.-O. (1978a). Bull. Soc. Chim. Belg. 87, 229-238.
Scheibye, S., Pedersen, B. S. \& Lawesson, S.-O. (1978b). Bull. Soc. Chim. Belg. 87. Submitted.


[^0]:    * Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 33796 ( 13 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH 1 2HU, England.

